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1 **Municipal leachates health risks: chemical and cytotoxicity assessment from**
2 **regulated and unregulated municipal dumpsites in Lebanon**

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6

7 **Abstract**

8 The proper management of municipal waste is critical for resource recovery, sustainability and
9 health. Lebanon main approach for managing its municipal waste consisted of landfill disposal
10 with minimal recycling capacity. This approach contributed to exceeding the holding capacity of
11 existing landfills leading eventually to their closures. The closure of a major landfill (Naameh
12 landfill) servicing Beirut and Mount Lebanon areas led to municipal wastes piling in the streets
13 and forests for more than a year in 2016. The main problem identified in the municipal wastes
14 consisted of untreated leachates (from regulated and unregulated dumpsites) going straight into
15 the Mediterranean Sea. Therefore leachate samples were collected and subjected to chemical
16 characterization followed by biological assessment. The chemical characterization and profiling
17 of the Lebanese leachates were compared to results reported in Lebanon, Europe and United States
18 as well as to the toxicity reference values (TRV). The biological assessment was conducted *in vitro*
19 using human derived immortalized cell cultures. This strategy revealed significant alarming
20 cellular organelles and DNA damages using *in vitro* cytotoxicity assays (MTS and comet assay).
21 The significant damages observed at the cellular level prompted further animal model
22 investigations using BALB/c mice. The animal data pointed to significant upregulation of liver
23 activity enzymes coupled with significant damage expression in liver spleen and bone marrow

24 DNA. The presented research clearly indicated that there is an urgent need for development of
25 national waste strategies for proper treatment and disposal of municipal waste leachates in
26 Lebanon.

27

28 *Keywords: Municipal waste, Leachates, toxicity, chemical characterization, in vitro assays, health*
29 *impacts*

30

31 **1 – Introduction**

32 Municipal waste disposal into landfills is the primary option adopted by numerous countries
33 worldwide (Baderna et al., 2011) (Mukherjee et al., 2015) (Gong et al., 2014). This disposal
34 method poses numerous challenges due to the unplanned development of landfill areas lacking
35 adequate engineering controls or even improper oversaturation of existing landfills. This leads to
36 the potential formation of toxic gases and leachates that can escape containment and find their way
37 to the soil and groundwater (Mukherjee et al., 2015).

38 Leachates consists of the liquid effluent generated from municipal waste consolidation into
39 landfills. This liquid is a cocktail of numerous chemicals that are the result of water passing
40 through the waste and saturating it with organic and inorganic matter. The produced leachate poses
41 significant disposal challenges for landfill operators worldwide due to its potential to contaminate
42 soils, surface water, aquifers and sea water.

43 Leachate composition is affected by numerous factors as outlined in Johansen and Carlson 1976
44 (Johansen and Carlson, 1976). Briefly these factors include landfill age, the geological conditions
45 present in addition to local weather affecting the hydrogeological conditions in the landfills. Other
46 important parameters to be considered within the landfill includes interaction of various chemicals,

47 the internal temperature and pH. Landfill consolidated breakdown can occur under aerobic and
48 anaerobic conditions. This breakdown contributes to stabilizing the organic component leading to
49 lower leachates organic and inorganic concentrations (Jędrzejak A. . 1994).

50 Landfill operators worldwide tend to compact the waste into landfills. This practice creates
51 anaerobic conditions which results in methane gas formation. The formation of methane is
52 observed in old landfills and is usually the result of acidogenic in young landfills followed by
53 methanogenic reactions in old landfills. It is also important to note that the composition of the
54 leachates together with climate and technology used in compacting the waste can play an important
55 role in altering the leachates profile (Slomczyńska and Slomczyński, 2004).

56 Many studies were conducted on landfill sites worldwide with a focus on mutagenicity (Deguchi
57 et al., 2007) and toxicity mainly in plants (Bhat et al., 2016). Across northern Spain for example,
58 four municipal solid waste landfill sites have been monitored for eleven perfluoroalkyl
59 carboxylates (PFCAs) and five perfluoroalkyl sulfonates (PFSA). PFSA's importance stems from
60 their C-F bonds making them highly stable in long alkyl chains (Prevedouros et al., 2006). This
61 property led to their extensive production and various applications including coating materials,
62 water repellent surfactants and fire retardant. (Busch et al., 2010) (Dauchy et al., 2012) (Yan et al.,
63 2015). PFSA's high stability makes them non-biodegradable, persistent and extremely difficult to
64 remediate using non-conventional methodologies (Quiñones and Snyder, 2009) such as membrane
65 bioreactors (Fotakis and Timbrell) with limited remediation success (Fuertes et al., 2017).

66 The cytotoxicity and DNA damage induction in four simulated landfill soil leachates from Nigeria
67 and India were also evaluated. These assessments were conducted using the MTT cell
68 proliferation assay for cell number determination assay and alkaline comet assay for the DNA
69 damage assessment (Swati et al., 2017). Heavy metals (Cadmium, Iron and Zinc) (Ratzinger et

70 al.), Polycyclic Aromatic hydrocarbons (PAH), polycyclic chlorinated biphenyls (PCBs) and
71 organic chemicals detected in samples were higher than allowable exposure limits. The researchers
72 also reported significant cytotoxic and DNA damage induction in exposed cells leading to
73 significant morphological alterations and apoptosis. All these results clearly indicated the
74 significant health risks posed by leachates exposure (Alimba et al., 2016). The assessment of
75 leachates from India also showed a similar pattern with reports of high level of organics (158 times
76 allowable limits) and lower heavy metal content. The paper highlighted the significant health risks
77 posed by low concentrations of PAH especially when these PAH interact synergistically to cause
78 cytotoxic and genotoxic damages (Ghosh et al., 2015).

79 Leachates from landfills in Turkey were also assessed. Researchers reported varying pH (4-8),
80 temperatures (2.8 and 24.5 °C) and organic compounds (33 in all) such as phthalates and
81 naphthalene and alkanes (Banar et al., 2006).

82 Lebanon leachates were previously analyzed and reported in a 2002 study by El-Fadel et al., (El-
83 Fadel et al., 2002). The leachates were analyzed for a number of parameters such as pH, COD,
84 Total organic Carbon (TOC), Total Dissolved Solid (TDS), chlorides, sulfates, orthophosphates,
85 nitrates, ammonia nitrogen, hardness, and heavy metals.

86 Recently in 2016, municipal waste has been an emerging concern in our Lebanon. The regulated
87 dumping sites could no longer manage the tremendous quantity of municipal garbage generated
88 and therefore numerous unregulated dumping sites were created throughout the country. This large
89 number of unregulated waste disposal areas could be correlated to environmental pollution and
90 human diseases. To minimize the impact of dumpsites on human health and the environment, a
91 qualitative and quantitative research into leachate production, toxicity and potential management
92 was needed. It is important to note that waste leachate was typically released to the environment

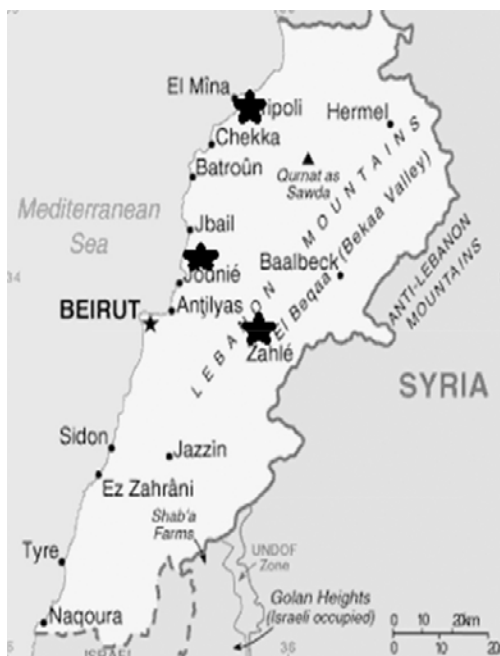
93 without any treatment, increasing the risk of environmental and human damages. Consequently,
94 we must improve our knowledge of such matrices to be able to develop reliable treatment
95 processes.

96 The main challenges for leachates management were mainly toxicological in nature (Kalka, 2012)
97 (Ghosh et al., 2017) (Slomczyńska and Slomczyński, 2004). The leachates chemical composition
98 was anticipated to vary between different dumping sites depending on the nature of the waste as
99 well as the climate. Establishing toxicity profile predictions for these leachates is challenging due
100 to the unique geochemical nature of each landfill and the variation in soil layers and water table
101 (Koshy et al., 2007).

102 The uncontrolled disposal of these leachates into the soil and waterways poses significant
103 challenges due to the composition of these leachates mainly in the form of organic and inorganic
104 contaminants (Nagarajan et al., 2012) (Raghab et al., 2013). Leachates also poses pressures on
105 Biochemical and Chemical Oxygen Demand (BOD and COD), TOC, ammonium and sulfur
106 compositions and heavy metals in soil and groundwater (Gajski et al., 2012). The mixture of
107 compounds generated by landfill is complex and more than 200 compounds of hazardous nature
108 have been previously identified in landfill leachates. These compounds ranges from aromatic to
109 phenols, halogenated and other compounds as described in the literature (Öman and Junestedt,
110 2008) (Baun and Christensen, 2004) (Adar and Bilgili, 2015).

111 The various chemical compounds in municipal leachates can lead to significant damages in
112 ecological systems, food chains and ultimately human population. These effects can range from
113 toxicity to carcinogenicity as reported by many researchers (Mukherjee et al., 2015) (Moraes and
114 Bertazzoli, 2005) (Gajski et al., 2012).

115 A map showing the municipal dumps leachates collection sites can be referred to in Figure 1.



117

118 **Figure 1:** Map of Lebanon indicating leachates sampling sites marked as * on the map.

119

120 The main motivation behind the research was to highlight the significant health and environmental
 121 risks posed by disposal of untreated municipal leachates into the natural environment. We
 122 anticipate that the findings will guide decision makers in the country to adopt better waste
 123 management policies and strategies. The novelty of this study is at two levels. Firstly, the research
 124 aims at establishing a linkage between the Lebanese leachates chemical compositions and the
 125 biological impacts observed at the cellular and animal levels (not previously reported in the
 126 literature). Secondly, we could not find any published data on the genotoxic impacts of Lebanese
 127 municipal leachates and this is a very important salient matter to address. The leachates genotoxic
 128 data clearly indicated the urgent need for the government to address the issue due to the long-term
 129 public health implications. This a very important matter because leachates exposure could
 130 potentially lead to future cancer onsets in the Lebanese population.

131

132 **2. Materials and methods**

133 This study evaluated the leachates sampled from a number of regulated (Tripoli and Zahle) and
134 unregulated municipal dumpsites (Ghazeer, Mount Lebanon) throughout Lebanon (El-Fadel, Bou-
135 Zeid et al. 2002). The five samples were collected from three sites as shown in Figure 1 and
136 consisted of the following:

- 137 1. Zahle old waste leachate, untreated from weathering pond next to landfill (sample A);
- 138 2. Zahle new waste leachate, untreated straight from landfill (sample B);
- 139 3. Tripoli waste leachates straight from Landfill, untreated (sample C);
- 140 4. Tripoli waste leachates biologically treated (sample D); and
- 141 5. Ghazeer waste leachates, unregulated municipal dumpsite untreated (sample E).

142 It is important to note that the selected sites didn't have any biological treatment except for Tripoli
143 landfill. The study consisted in characterization of chemical composition of landfill leachates. The
144 analytical assessment was followed by *in vitro* (cellular model) and *in vivo* (mice model) to
145 establish the biological impacts of leachates exposures. It is important to note that the fate of
146 treated and untreated landfill leachates in Lebanon was ultimately the Mediterranean Sea. The
147 untreated leachates originated predominantly from inland and sea front landfills trucked in using
148 major highways prior to disposal into the sea. Sea disposal was through a 1.5 km pipeline in an
149 area categorized as dead zone known as Ghadir. This area was used for disposal of industrial and
150 other waste discharges over many years.

151 *2.1. Study design*

152 Leachates were collected in glass bottles on ice prior to transportation to the labs using the standard
153 method for leachate examination (American Public Health et al., 1998). Samples were processed
154 for chemical characterization, and cytotoxicity assessment. The samples used for cytotoxicity

155 assessment were centrifuged at high speed to remove any particles and then filter sterilized using
156 a 0.2 micrometer filters. The samples were then diluted prior to conducting the cytotoxicity and
157 animal toxicity assessments.

158

159 *2.2. Analytical assessment*

160 Bis(2-ethylhexyl) phthalate, pestanal analytical standard was purchased from Sigma-Aldrich. The
161 metal standards (1000 mg/L in 0.5 M HNO₃) were purchased from Merck whereas anions were
162 purchased from Ultra Scientific. All HPLC/GC grade solvents and other chemicals were purchased
163 from Sigma-Aldrich or Fisher Scientific and used without further purification.

164 *2.2.1. Preparation of waste leachates organic compounds*

165 The organic compounds were extracted from 50 mL of leachates with dichloromethane (DCM)
166 solvent (500 mL, 3 times) using liquid-liquid extraction method. After extraction, the DCM
167 solution was dried using calcium sulfate, filtered then evaporated using rotary evaporator under a
168 pressure of 500 mBar and 40 °C. The resulting solid was reconstituted in 1 mL of DCM prior to
169 GC/MS analysis. The tested samples were filtered with Cronus Syringe Filter PTFE 25 mm 0.45
170 µm to remove any suspended particles.

171 *2.2.2. GC/MS analysis*

172 Samples from municipal leachates were subjected to GC/MS using a GC/MS (Hewlett Packard,
173 HP6890 fitted with a fused silica HP5-MS 5% phenyl methyl siloxane column (30 m x 0.25 mm)
174 i.d., film thickness (0.25) and coupled to MS detector. Helium was the bearer gas used with
175 splitless infusion (1 µL infusion volume) at a rate of 1.2 mL/min. The temperature program was
176 2.0 min at 70 °C, from 70 °C to 130 °C at 8 °C/min and hold for 5 min, from 130 °C to 180 °C at

177 2 °C/min and hold for 10 min, from 180 °C to 220 °C at 15 °C/min and hold for 2 min, from 220
178 °C to 280 °C at 15 °C/min and hold for 45 min and after that from 280 °C to 300 °C at 15 °C/min
179 and hold for 20 min. The assessment was conducted in scan mode to broaden the search process.
180 Compounds were identified using the NIST11 and W9 libraries. The presence of di(2-ethylhexyl)
181 phthalate (DEHP) in the five samples was confirmed by GC/MS using its commercial form as
182 standard, Figure S1-S2.

183 *2.2.3. Chemical characterization and statistical analysis*

184 Leachates samples were vacuum sucked through a 0.45 µm pore cellulose filters, divided into two
185 samples in polyethylene bottles. One sample was acidified with nitric acid to pH < 2 and stored at
186 4 °C for metal analysis (triplicates) (Pb, Cd, Cr, Cu, Al, Mn, Fe, Zn, As, K, Na and Ni) using AAS
187 Graphite Furnace technique (“Shimadzu” AAS-6300). The working standards were prepared by
188 stock solution dilution (1 mg metal/mL in 2% HNO₃) with MilliQ water. The second sample was
189 used for determination of a number of parameters such as pH, conductivity (Hach Model 44600
190 Conductivity/TDS Meter, resolution Conductivity 0.1 µS/cm), turbidity (Nephelometric method),
191 titration of Ca, Mg, total hardness (0.01 M EDTA), spectrophotometry for NO₃⁻ (cadmium
192 reduction), SO₄²⁻ (turbidimetry), PO₄³⁻ (ascorbic acid), TDS, FS (fixed solid) and VS (volatile
193 solid) using LOI (loss on ignition). The analysis were conducted with certified prepared reagents
194 (EPA standards) obtained from Hach Chemical Company, USA. All procedures adopted were in
195 accordance with the standard method of water analysis and waste water (APHA, AWWA, WPCF
196 (2012)). Chloride Cl⁻ was determined by ion chromatography (Ion Chromatograph, “Shimadzu”
197 PIA-100). TC (total carbon), TOC using a TC and TOC Analyzer “Shimadzu” VCPN.

198

199 *2.3. Cytotoxicity assessment*

200

201 *2.3.1. Human modified keratinocyte cells (HaCaT)*

202 Human HaCaT keratinocytes (non-tumorigenic) were derived from normal keratinization in a
203 spontaneously immortalized aneuploid human keratinocyte cell line (Boukamp et al., 1988). The
204 cells were selected as they represented the first barrier in the skin in contact with leachates.

205

206 *2.3.2. Cell culture protocols*

207 Cells were maintained at 37 °C in a 5% CO₂ incubator. The cell culture medium used consisted of
208 DMEM (Dulbecco's Modified Essential Medium; Gibco) with 5% fetal calf serum, L-glutamine
209 (2 mM), penicillin (100 U/mL), streptomycin (0.1 mg/mL; Sigma, USA). Upon confluence cells
210 were subjected to trypsinization and seeded at a density of 5 x 10³ cells/mL. Cells viability
211 surpassed 95% as shown by Tryptan Blue (Sigma, USA) analysis. The 24h post exposure
212 assessment was conducted by further incubating exposed cells in fresh media for 24 h prior to
213 assessment.

214

215 *2.3.3. In vitro cytotoxicity assays*

216

217 *2.3.3.1. MTS assay*

218 Cell mitochondrial dehydrogenase activity was undertaken using the MTS assay kit from Promega
219 as previously described (Khalil, 2015) (Khalil and Winder, 2008).

220

221 *2.3.3.2. Comet assay*

222 The assay investigated the single stand and double strand DNA damages. The methodology used
223 has been extensively described (Khalil and Shebawy, 2017).

224 *2.4. Cell experimentation statistical analysis*

225 All data presented in this paper represented the mean and standard deviation of three independent
226 experiments with at least three replicates. Statistical analysis were conducted by Microsoft Excel,
227 and Graphpad. Statistical significance was reported for $p < 0.05$.

228

229 *2.5. Mice model assessment*

230 The BALB/c mice model was selected as the biological indicator for assessment. Briefly mice
231 were kept in cages in groups of five and maintained under a 12 h. photoperiod (08:00–20:00) at an
232 environmental temperature of $22\text{ }^{\circ}\text{C} \pm 2\text{ }^{\circ}\text{C}$. The mice were deprived of normal drinking water and
233 were provided instead with undiluted/diluted leachates as their main water source. The mice were
234 exposed to leachates samples coming from the dumpsites where leachate treatment (microbial or
235 environmental attenuation) was undertaken. The mice drank the leachate water for a period of 4
236 weeks prior to being sacrificed and analyzed for organ damages. During the trial period mice were
237 weighed daily. All experimental protocols were approved by the Animal Ethical Committee of the
238 Lebanese American University, which complies with the Guide for the Care and Use of Laboratory
239 Animals (Committee for the Update of the Guide for the Care and Use of Laboratory Animals,
240 2010). Blood was also collected from every mouse within each group directly from the inferior
241 vena cava using a 5 mL syringe containing Na_2EDTA (1 mg/mL), and transferred to 5 mL tubes
242 for analysis.

243 The samples were kept on ice and then were centrifuged at 3000 rpm for 30 mins. The supernatants
244 were collected and tested for the alkaline phosphatase (ALP), glutamic-oxaloacetic acid
245 transaminase (GOT) and glutamic-pyruvic acid transaminase (GPT) enzymes using standard
246 commercial Spinreact kits.

247 Mice were sacrificed at the end of the exposure period and liver, spleen and kidneys were
248 harvested. The organs were weighed on a calibrated balance and the results were recorded. Bone
249 marrow cells were collected from animal carcasses using the method described in (Swamydas and
250 Lionakis, 2013). Euthanized mice were subjected to surgery to remove the lower extremities. The
251 femur and tibia were cleaned from any muscles and sprayed with 70% ethanol. The bones were
252 placed in a Petri dish containing ice-cold RPMI 1640 1X supplemented with 10% FBS and 1%
253 Penicillin/streptomycin. Bones were washed in ice cold PBS and broken in a sterile environment.
254 Using a 12 cc syringe filled with RPMI supplemented with 10% FBS and 2 mM EDTA, the bone
255 marrow was flushed out before being analyzed using the comet assay methodology.
256 The organs were also sliced and subjected to digestion in trypsin overnight at 4 °C in a shaker.
257 Cells from each organ were harvested the next morning though pipetting and subjected to the
258 comet assay for DNA damage determination.

259

260 **3. Results**

261 *3.1. Analytical chemistry*

262 Five samples A, B, C, D and E were analyzed using GC/MS without dilution nor filtration prior to
263 extraction. This ensured that no molecules were removed by pre-treatment and most information
264 associated with the sample was retained. The extract was then filtered with 0.45 µm filter to
265 remove any suspended particles prior to GC/MS injection. GC/MS was conducted on waste
266 leachate organic composition (for the five samples) and chromatograms obtained. All the
267 compounds listed (Table 1) in this part were identified using the NIST11 library, and matches
268 between the experimental mass spectrum and theoretical ones were over 90% for every compound.

269 Compounds that were not confidently identified by the software did not exceed 3% by area of any
 270 sample mixture and were excluded from Table 1 for clarity.

271 **Table 1:** List of the compounds identified by GC/MS in samples A, B, C, D, and E. The relative
 272 integrated area percentage for each identified compound is tabulated and indicates the relative
 273 abundance of each compound in the extracted mixture.
 274

Peaks number	Area %	Compound
Sample A		
1	56.84	Oleamide
2	30.66	Di(2-ethylhexyl) phthalate (DEHP)
3	12.51	Hentriacontane
Sample B		
1	29.42	Cyclic octaatomic sulfur
2	21.43	Oleamide
3	2.68	Hexanedioic acid, bis (2-ethylhexyl)ester also known as DEHA
4	43.89	DEHP
Sample C		
1	100	DEHP
Sample D		
1	68.27	Tributylamine (TBA)
2	13.74	Oleamide
3	18.00	DEHP
Sample E		
1	5.69	Nicotine
2	6.55	Hexanedioic acid, bis (2-ethylhexyl) ester also known as DEHA
3	2.35	Pentacosane
4	78.80	DEHP
5	5.69	Eicosane

275 The list of organic compounds extracted from the samples is not comprehensive. The heterogeneity
 276 of the original samples may be hindering the extraction of persistent organic pollutants especially
 277 the ones possessing low concentration in the aqueous layer. Other persistent organic pollutants
 278 (POPs) including PAHs, polychlorinated dibenzofurans (PCDFs), polychlorinated dibenzo-p-
 279 dioxins (PCDDs), polychlorinated dibenzothiophenes (PCDTs), polybrominated diphenyl ethers
 280

281 (PBDEs) and PFASs which were detected in reported leachates samples, were typically
282 undergoing a different extraction and detection procedures to the one adopted in this study (Gallen
283 et al., 2017) (Dudzinska M R, 2011). DEHP which was not detected in the blank sample (Figure
284 S1-S2), was common compound in all samples.

285
286 Advanced chemical characterization was also performed on leachate samples of the five sites and
287 several metals were detected including Pb, Cd, Cr, Cu, Al, Mn, Cu, Zn and As (Figure 2).

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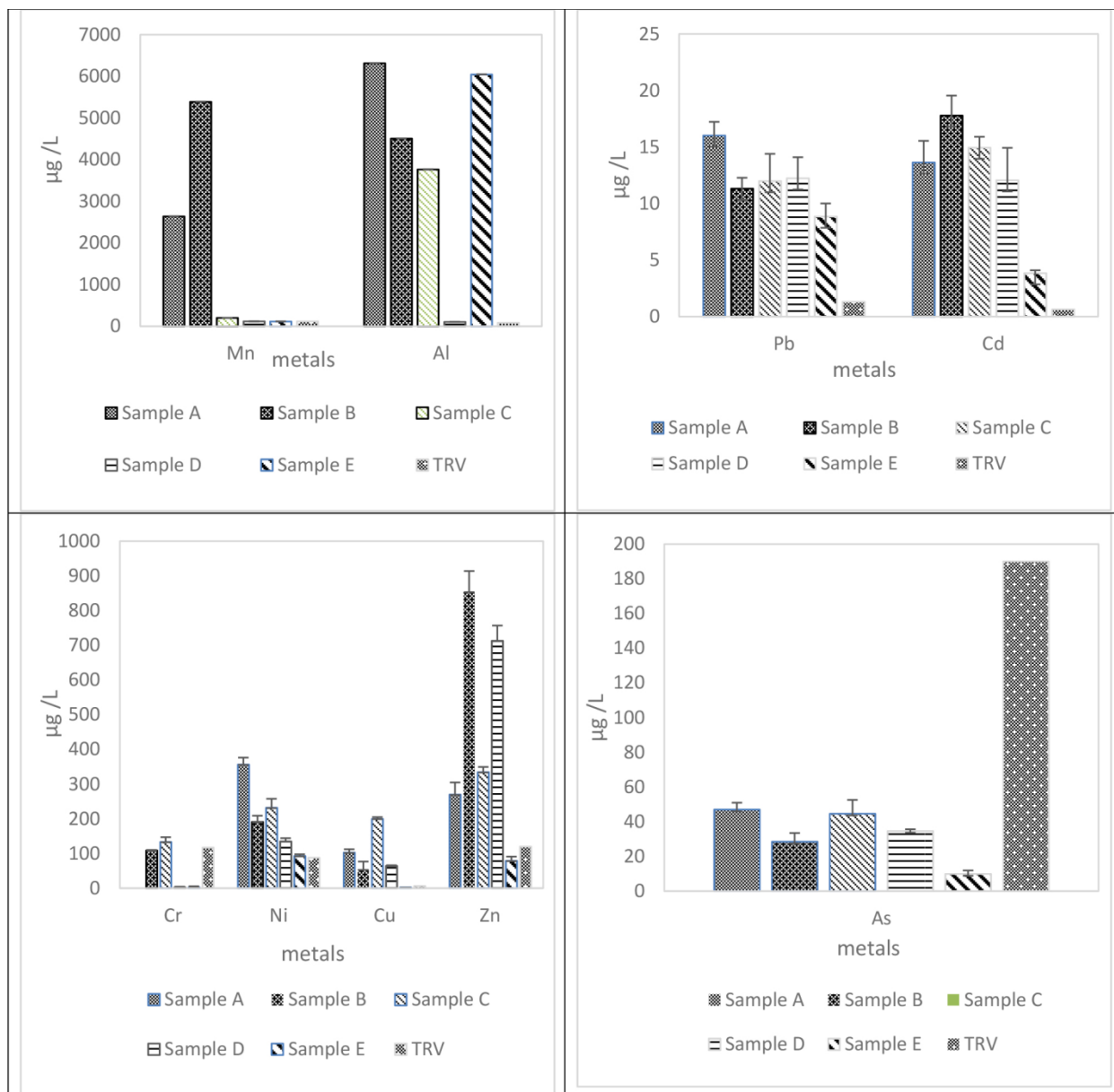


Figure 2: Variation of toxic metals among different samples with data related to TRV values.

297
 298
 299
 300
 301 Other important parameters were also evaluated such as K, Na, and Ni, NO_3^- , SO_4^{2-} , PO_4^{3-} ,
 302 hardness, Cl^- , Ca, Mg, conductivity, turbidity, TOC, TC, IC, TDS, FS and VS (Table 2). The ranges
 303 of these parameters in all the five sites were calculated and compared to previously reported data
 304 in Lebanon (in year 2002), Norway and USA.

305
306
307

Table 2: Compositions of leachates from the five landfill sites compared to Lebanon (El-Fadel et al., 2002), Norway (Slomczyńska and Slomczyński, 2004) and USA (Johansen and Carlson, 1976)

Parameters	Units	Sample A	Sample B	Sample C	Sample D	Sample E	Range of samples A→E	TRV Unit: ppb = µg/L	Lebanon Unit: mg/L	Norway Unit: mg/L	USA Unit: mg/L
pH		7.85	7.44	8.12	6.34	5.81	5.81 - 8.12	N/A	1.5 - 9.5	5.9 - 7	5.4 - 6.4
Color true	PtCo unit	21800	26300	9875	600	7500	600 - 26300	N/A	N/A	N/A	N/A
Lead (Georgie J et al.)	µg/L	16.10	11.33	12.00	12.25	8.86	8.86 - 16.03	1.32	0 - 14.2	0.001 - 0.015	0.1 - 1.40
Cadmium (Cd)	µg/L	13.65	17.80	14.95	12.08	3.86	3.86 - 17.80	0.66	0 - 1.16	0.0001 - 0.002	0.01 - 0.03
Chromium (Cr)	µg/L	-	10.88	133.33	22.30	24.50	2.23 - 133.33	117.32	0 - 22.5	0.002 - 0.17	0.05 - 1.05
Copper (Cu)	µg/L	102.50	53.85	199.80	64.28	2.47	2.47 - 199.80	6.54	0 - 9.9	0.008 - 0.085	0.18 - 1.30
Aluminum (Al)	µg/L	6316	4507	3767	107	6046	107 - 6316	87	0.5 - 85.0	N/A	N/A
Manganese (Mn)	µg/L	2640	5389	205	120	117	117 - 5389	120	0.05 - 1400	N/A	N/A
Iron (Fe)	µg/L	9634	9543	8451	1741	4692	1741 - 9634	1000	0 - 42000	11.5 - 234	245 - 810
Zinc (Zn)	µg/L	270	854	334	713	79	79 - 854	120	0 - 1000	0.055 - 2.65	5.3 - 155.0
Arsenic (As)	µg/L	47.00	28.48	44.69	34.65	9.87	9.87 - 46.98	190	0 - 70.2	N/A	N/A
Potassium (K)	µg/L	9162	8945	9805	8568	3704	3704 - 9805	53000	0.16 - 3370	21.3 - 219	N/A
Sodium (Na)	µg/L	9161	8945	9804	8567	3704	3704 - 9804	680000	0 - 8000	34.8 - 462	N/A
Nickel (Ni)	µg/L	356.50	191.00	231.80	135.00	92.70	92.68 - 356.50	87.71	0 - 7.5	0.005 - 0.12	0.10 - 1.20

Parameters	Units	Sample A	Sample B	Sample C	Sample D	Sample E	Range of samples A→E	Lebanon Unit: mg/L	Norway Unit: mg/L	
Nitrate - (NO ₃ ⁻)	mg/L	2	8	2	101	23	2 - 101	0 -9.8	N/A	
Phosphate- (PO ₄ ³⁻)	mg/L	41.00	50.25	32.65	8.00	48.10	8 00- 50.25	0.01 - 154	N/A	
Sulfate- (SO ₄ ²⁻)	mg/L	6	100	525	20	1525	6 - 1525	0 - 1850	10 - 100	
Chlorides- (Cl)	mg/L	650	775	500	115	650	115 - 775	11375	68 - 680	
Calcium- (Ca)	mg/L	62062	94094	12012	30030	230230	12012 - 230230	5 - 4080	99 - 400	
Magnesium- (Aliahmadi et al.)	mg/L	10010	29136	7284	12140	18210	7284 - 29136	0 -115600	13 - 96	
Water hardness	Total	mg/L	180000	60000	125000	650000	355000	60000 - 650000	N/A	N/A
	Calcium	mg/L	155000	30000	75000	575000	235000	30000 - 575000	(CaCO ₃) 0.1 -225000	N/A
	Magnesium	mg/L	25000	30000	50000	75000	120000	50000 - 120000	N/A	N/A
Conductivity	µs/cm	19260	23400	40300	1706	36000	1706 - 40300	480 -72500 (µs/cm)	N/A	
Turbidity	NTU	320	310	96.25	36.24	1184	36.24 - 1184	40 - 500	N/A	
TOC- (Total Organic Carbon)	mg/L	878	990	278	2	506	2 - 1184	335000	30 - 1700	
TC- (Total Carbon)	mg/L	1415	1510	602	45	621	45 - 1510	N/A	N/A	
IC- (Inorganic Carbon)	mg/L	537	500	304	43	98	43 - 537	N/A	N/A	
TDS -(Total Dissolved Solid)	mg/L	26730	34700	12350	1100	17520	1100 - 34700	584 -55000	N/A	
FS- (Fixed Solid)	mg/L	12690	16050	8030	600	10310	600 - 16050	N/A	N/A	
VS- (Volatile Solid)	mg/L	14040	18650	3320	500	7210	500 - 18650	N/A	N/A	

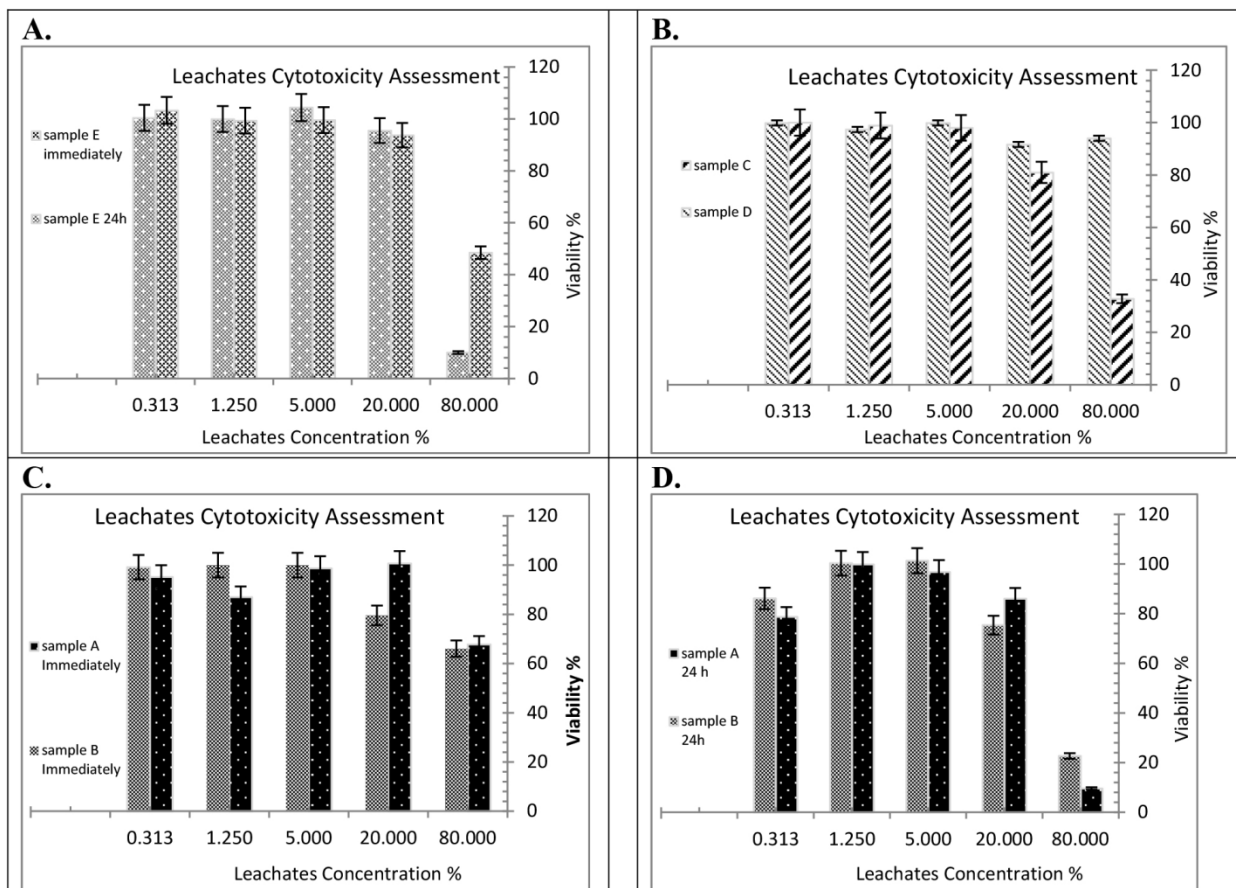
308
309

310 3.2. Cytotoxicity results

311 3.2.1. Leachates exposure

312 The viability of cells exposed to leachates was assessed using the MTS assay. Cells were treated
313 with different filter sterilized dilution of raw leachates for 2 h. (immediately) and 24 h. from
314 samples A, B, C, D and E and cytotoxicity assessment of leachates were evaluated using the MTS
315 assay (Figure 3).

316



317 **Figure 3:** Leachate Cytotoxicity assessment from various locations throughout Lebanon.

318 The cytotoxicity results from the MTS assay and the expression of significant toxicity at 80%
319 dilution immediately and 24 h. post exposure warranted the investigation of potential DNA
320 damages triggered by leachates exposure. The aim of this approach was to determine whether a
321
322

323 similar profile to cytotoxicity analysis could be observed in DNA damage triggered by leachates
 324 exposure.

325

326 *3.2.2. Leachates DNA damaging potential on cell cultures using comet assay*

327 The DNA damage for the five samples was assessed using alkaline comet assay (Table 3). The
 328 DNA damage in cells exposed to leachates was shown using seven parameters such as head length,
 329 tail length, comet length, head DNA content, tail DNA content and tail moment.

330

331 **Table 3:** DNA damage analysis of HaCaT cells exposed to leachates. This was measured by the
 332 comet assay immediately (4h) post exposure with CASP (Comet assay Software package).
 333

	Positive control	Negative control	Sample A	Sample B	Sample C	Sample D	Sample E
Head length	75.66 ± 9.68	115 ± 12.16	81 ± 20.13	119 ± 30.02	67 ± 3.05	81 ± 11.01	87 ± 10.06
Tail length	97.66 ± 12.11	4 ± 1	129 ± 11.01	135.6 ± 39.33	92.66 ± 4.33	69.66 ± 7.17	106.6 ± 5.54
Comet length	173.3 ± 18.02	119 ± 12.28	210 ± 24.84	254.6 ± 68.67	159.66 ± 6.74	150.66 ± 5.66	193.6 ± 12.70
Head DNA content	27.07 ± 5.76	99.63 ± 0.19	26.33 ± 5.32	43.67 ± 5.58	26.20 ± 1.03	45.44 ± 8.27	30.05 ± 4.67
Tail DNA content	72.92 ± 5.76	0.36 ± 0.19	73.66 ± 5.32	56.32 ± 5.58	73.79 ± 1.03	54.55 ± 8.27	69.94 ± 4.67
Tail moment	72.51 ± 13.95	0.01 ± 0.006	94.12 ± 5.23	73.21 ± 15.33	68.40 ± 3.48	39.16 ± 10.18	74.44 ± 5.13
Overall tail moment	45.56 ± 9.03	0.18 ± 0.08	60.27 ± 4.12	56.7 ± 10.01	44.56 ± 1.62	29.67 ± 3.90	59.20 ± 12.12

334

335

336 The tail moment Index (TMI) was calculated to reflect DNA damage in exposed HaCaT cells
 337 (Figure 4).

338

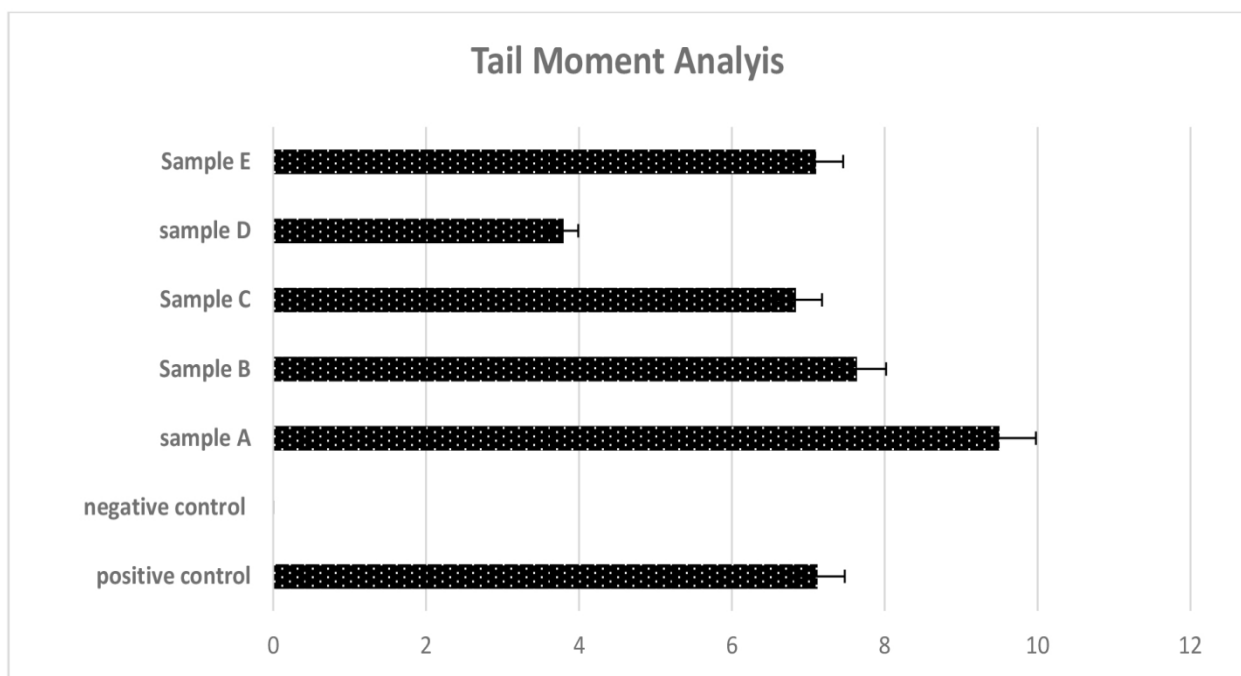


Figure 4: Tail moment analysis of comet assay data from leachate samples.

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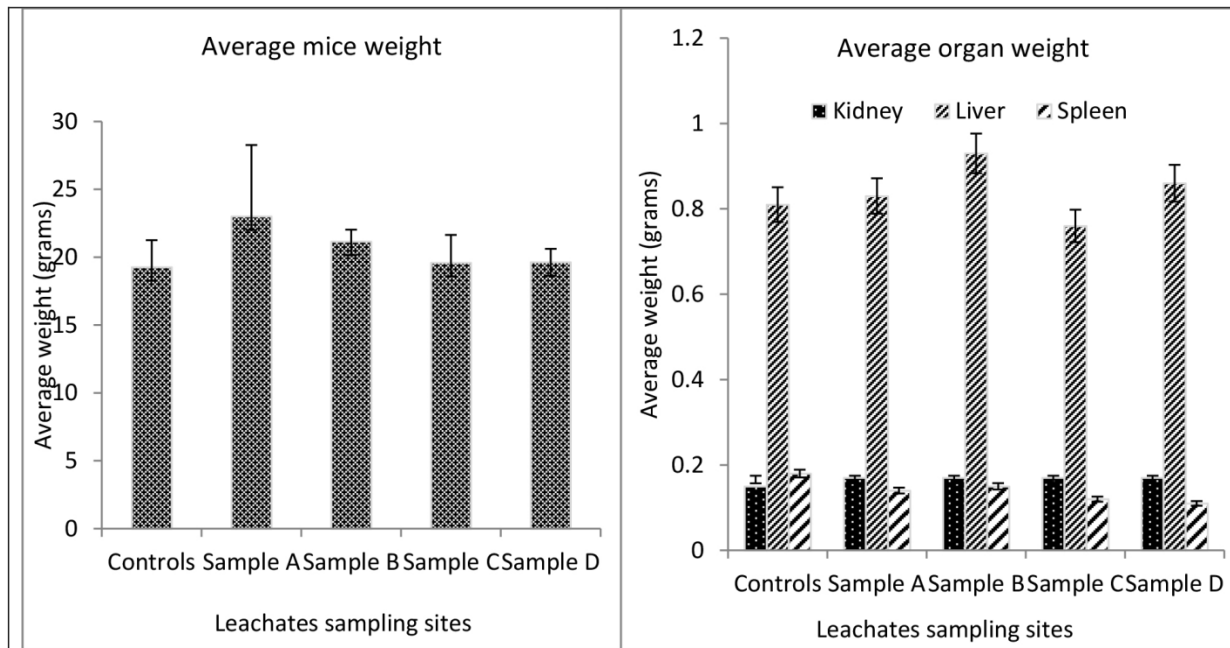
342 3.3. Leachates toxicity assessment using mice model

343 Mice were exposed to leachates as previously described in the material and method section.

344 Average mouse overall weight in exposed and control mice was reported in Figure 5a. In controls
345 the average weight was 19 grams, in sample A 23 grams, in sample B 21 grams, in sample C 19
346 grams, and in sample D 19 grams.

347 The organs of the mice were also harvested, and the weight of liver, kidney and spleen were also
348 reported in Figure 5b.

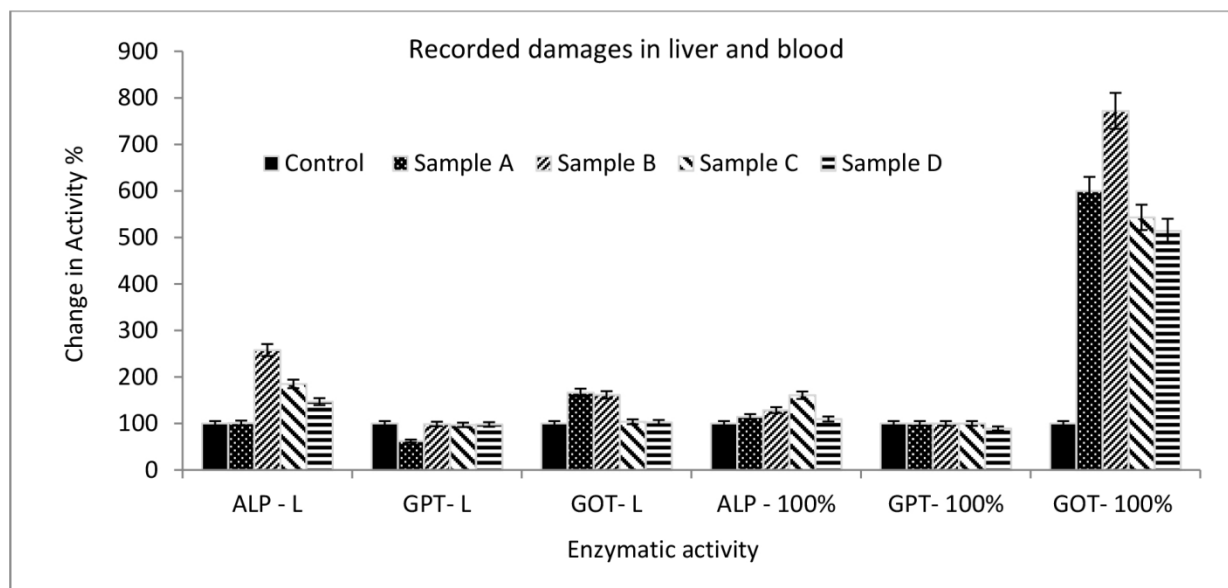
349



350 **Figure 5:** Mice overall and organ weights upon leachates exposure. The liver and blood of
 351 sacrificed mice was collected and subjected to several tests to establish enzymatic activities (Figure
 352 6).
 353

354 The ALP, GOT and GPT enzymes activities of the blood (ALP, GOT and GPT-100%) and the
 355 liver (ALP, GOT and GPT-L) of exposed and control mice was reported in Figure 6. We recorded
 356 significant increases in GOT levels upon leachates exposures in all samples.
 357

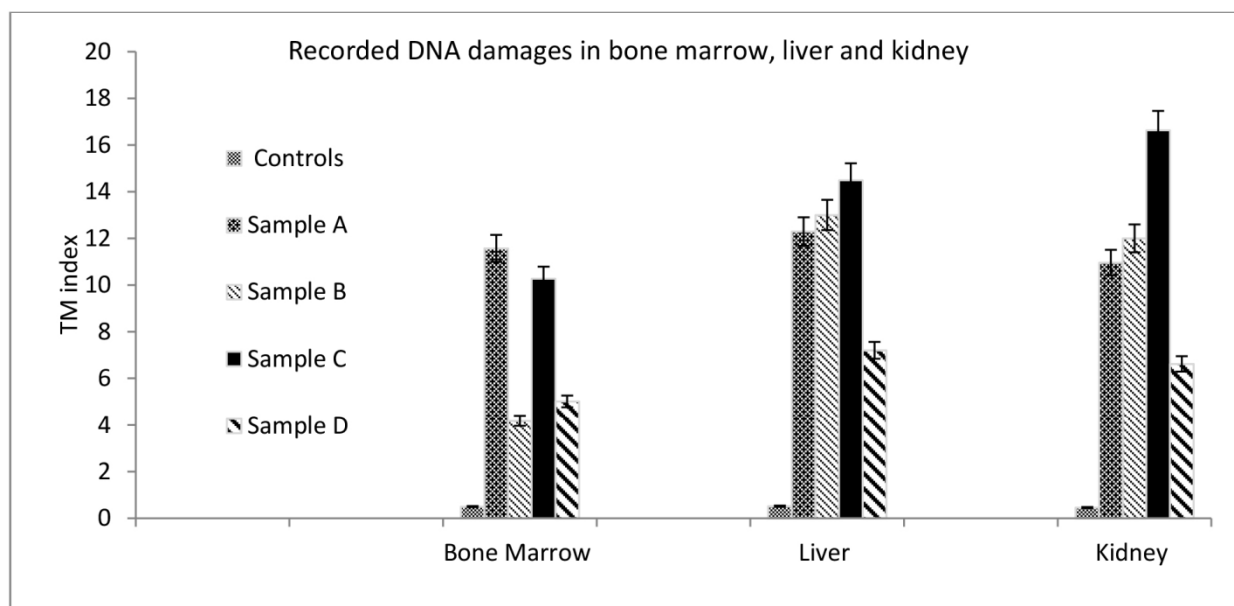
358



359 **Figure 6:** Enzymatic activity in blood of treated and untreated mice
 360

361
362 The DNA damaging potential of leachates drinking was assessed by harvesting the DNA from
363 isolated bone marrow, liver and kidney cells from exposed and control animals. The calculated
364 TM index (details in section 3.2.2) was reported in Figure 7.

365



366
367 **Figure 7:** Tail moment index indicative of DNA damages in treated and untreated mice
368

369
370 **4- Discussion**

371 *4.1. Leachates chemical characterization*

372 The chemical characterization of the leachate samples led to identification of numerous
373 compounds that were absent from the blank chromatogram as presented in Table 1, Figure S2.
374 This blank was performed following the same extraction procedures but excluding the leachate
375 sample presence, Figure S2. Leachate samples from different dumpsites had common chemicals.
376 Among the identified chemicals oleamide, identified in samples A, B and D is reported as a sleep-
377 inducing amidated lipid found in the cerebrospinal fluid of sleep-deprived cats (Cravatt et al.,
378 1995) (Gobbi et al., 1999) in addition to its use as a slip agent in polyethylene filaments (Garrido-

379 López et al., 2006). Another common compounds were n-alkanes (hentriacontane and eicosane)
380 mainly identified in samples A and E. Eicosane is commonly used in cosmetic, lubricants and
381 plasticizers (Larranaga, 2016). Solid n-eicosane (paraffin waxes) are also used as feeds for
382 cracking gasoline in addition to its use in oxidation, and chlorination cracking reactions (Schmidt,
383 2016). Eicosane was also reported as a principal component of diesel exhaust nanoparticles (Kanno
384 et al., 2008). Other identified chemical compounds such as Hexanedioic acid, bis(2-
385 ethylhexyl)ester or DEHA (in samples B and E) is used in the form of DEHP in medical products,
386 children's products, and plastic cling wrap for food storage (Rodgers et al., 2014) (Testai et al.,
387 2016), DEHA triggers slight irritation (in rabbit) and shows mild acute toxicity (Van Vliet et al.,
388 2011) and is classified Category 3 carcinogen by the International Agency of Research on Cancer
389 (Cancer) (Van Vliet et al., 2011) (Cancer, 2000). Tributylamine (TBA), identified in sample D, is
390 used as an inhibitor in hydraulic fluids; and chemical intermediate (Larranaga, 2016). DEHP
391 identified in samples A, B, C, D and E, is a hazardous plastic additive hazardous to human health
392 and reproductive system(Lithner et al., 2009) (Castillo and Barceló, 2001). DEHP is also present
393 in plastic products (wall coverings, furniture upholstery, garden hoses, automobile upholstery and
394 many other applications (ATSDR, 2002). The non-occupational exposure to phthalates poses high
395 risks due to its wide use in a wide range of consumable products. Post exposure, phthalates are
396 rapidly hydrolyzed to their respective monoesters, that is usually further bio transformed and
397 excreted in the urine and feces (Barr et al., 2003). Phthalates also displayed *in vitro* and *in vivo*
398 toxicity (mutagenicity, developmental toxicity and reproductive impairment among others (Bang
399 et al., 2011).

400 Some compounds were exclusively found in a single sample. For examples, cyclic octaatomic
401 sulfur, identified in sample B, is a microbiological activity indicator which is the result of organic

402 matter degradation in waste leachates. This could be the result of bacteria is using hydrogen sulfide
403 (H₂S) instead of water as an electron donor in a primitive photosynthesis-like process. The lack of
404 sulfur containing compound in other samples could be due attributed to its consumption during
405 Organic Matter (OM) degradation, whereas OM degradation is still occurring in sample B which
406 can be classified as freshly produced waste leachates (Badoil and Benanou, 2009). Nicotine,
407 identified in sample E, is a substance present in tobacco. Pentacosane, identified in sample E, is a
408 naturally occurring compound and constituent of the waxes (Peris-Vicente et al., 2006) (Belge et
409 al., 2014).

410 The only compound that exists in the five samples is DEHP. DEHP has wide applications as
411 plasticizer in PVC to make the final product soft and malleable. Plasticizers poses significant risks
412 due to their ability to migrate from the parent material and leach out into the environment. The
413 breakdown products of DHP have endocrine disruptors properties in addition to displaying higher
414 toxicity than DEHP itself. DEHP is resistant to microbial biodegradation due to the positioning of
415 the two ester groups to one another, as well as the branching on the side chains. The US EPA has
416 expressed significant concerns about phthalates due to their toxicity and environmental
417 risks(Erythropel et al., 2014).

418 Chemical analyses were performed on leachate samples collected from the five sites and several
419 parameters were monitored including pH, Pb, Cd, Cr, Cu, Al, Mn, Fe, Zn, As, K, Na, and Ni, NO₃⁻
420 , SO₄²⁻, PO₄³⁻, water hardness, Cl⁻, Ca, Mg, conductivity, turbidity, TOC, TC , IC (Inorganic
421 Carbon), TDS, FS and VS in Table 2. Due to the lack of standards data for metals values in
422 leachates, we compared our data to those cited in Lebanon in year 2002 (El-Fadel et al., 2002), in
423 Norway (Slomczyńska and Slomczyński, 2004) and USA (Johansen and Carlson, 1976) and the
424 aquatic TRVs for the Ecological Risk Assessment (ERA) (Division, 1999). Nitrate concentration

425 ranged between 2 mg/L and 101 mg/L with highest value measured in sample D (treated leachates)
426 which also showed the lowest levels of TOC (2 mg/L). Sample C (untreated waste sample) by
427 contrast had a high value of TOC (278 mg/L) and exhibited very low nitrate levels (2 mg/L). The
428 nitrate range was comparable to the levels previously reported in Lebanon except for sample D
429 (treated leachates) which was 10 times higher the reported values in 2002. This can be attributed
430 to nitrate actively oxidizing the water organic constituent in high TOC content, hence being present
431 in reduced form, whereas its oxidized form prevails when the TOC content is low such as in the
432 treated sample D (Zhang et al., 2016). The TOC range is within the Norway range and below the
433 one previously found in Lebanon. In the assessed samples phosphate levels ranged between 8 mg/L
434 and 50 mg/L, which are within the range of the previous study done in Lebanon. Water hardness,
435 as calcium and magnesium, were significantly higher than those previously reported. This could
436 be due higher alkalinity of the previously reported samples which renders dissolution of calcium
437 and magnesium carbonate more favorable and increases the total hardness. Figure 2 portrays the
438 variation of metals within sampling sites and includes the TRV values of aquatic organisms for
439 comparison. This assessment helps in projecting the long-range health risk potential of the
440 different components assessed here. As leachates are generated by solid waste land disposal sites,
441 contamination of ground and surface waters may occur through the process of bioaccumulation
442 and bio magnification. The health effects from leachate are not limited to drinking water as it can
443 penetrate skin, get into the food chain leading to metals accumulating in aquatic organisms. (James,
444 1977). From the analysis of the data of this study (Table 2 and Figures 2), 60 % of sampling sites
445 had Mn levels above those of TRV (120 µg/L), particularly sample B was higher by 44-fold; 80 %
446 of sites had Al levels far beyond the TRV value (87 µg/L) reaching at sampling site A levels higher
447 by 73-fold. In addition, 83 % of the samples had Cu levels higher than TRV (6.54 µg/L), reaching

448 at sample site C 30-fold higher, and 80 % of sampling sites had Zn levels higher than TRV (120
449 $\mu\text{g/L}$), with highest at site B by 7-fold. Nevertheless, all sites had higher levels of, Ni, Pb and Cd
450 than their TRV values of 87.71 $\mu\text{g/L}$, 1.32 $\mu\text{g/L}$, and 0.66 $\mu\text{g/L}$ respectively. Ni at site B was only
451 2-fold higher than the TRV value, while at site A Pb was 12-fold higher and Cd at site B was 27-
452 fold higher. Regarding As, all sites were lower than TRV (190 $\mu\text{g/L}$). The speciation of these
453 metals and their geochemistry was examined. The association of Cu with organic constituents was
454 highly evident upon comparing Cu levels of sample site C (199.8 $\mu\text{g/L}$) with high TOC and those
455 levels of Cu in sample D (64.28 $\mu\text{g/L}$) with low TOC. This can be attributed to the stability of Cu
456 organic complexes (Shaheen and Rinklebe, 2014). The levels of Cr were much higher in sample
457 C (untreated sample) than sample D (treated sample). A most probable explanation arises from the
458 reducing media of untreated sample C rendering Cr(III) a dominant form which along with basic
459 pH (8.12) leaves Cr as suspended $\text{Cr}(\text{OH})_3$ particles. It is important to mention here that all samples
460 were tested without filtration. An association of Cd with inorganic carbon (carbonates) was
461 evidenced in comparing sample C with high inorganic carbon (304 mg/L) and high Cd content
462 with those of site D (43 mg/L of carbonate) and low Cd content (Table 2). The strong association
463 of Cd to could be the result of similarities in the ionic radius of Cd (0.97 Å) and Ca (0.99Å). This
464 means that the Cd has the potential to enter the calcite crystal as camouflaged element and co-
465 precipitate with carbonate (Korfali and Jurdi, 2011) (Korfali, 2010).

466 The higher levels of Zn in treated sample (D) than untreated (C) can be explained by the association
467 of Zn with inorganic carbon (Korfali and Davies, 2004) in sample D which under acidic pH (pH=
468 6.34), the precipitated ZnCO_3 can readily dissolve and increase the level of Zn. To compare all
469 measured parameters among the different samples, the statistical t-paired test was performed using
470 the statistical SigmaStat Package. The statistical test was run to observe the parameters in samples.

471 Results indicated a statistical significant difference in measured parameters for sample C and D
472 (P= 0.006). Also, statistical significant differences existed between sample B and D (P <0.001), A
473 and D (P =0.024), E and D (P= 0.011).

474 The chemical analysis results discussed in the previous sections clearly indicated the presence of
475 many chemical and metal compounds from various origins in the municipal waste leachates. The
476 complex interaction between the various chemicals identified in the leachates makes it virtually
477 impossible to establish the true toxicity and genotoxicity of the leachate liquids from individual
478 chemicals detected. This complexity stems from the interactions between the different chemicals
479 and the potential additive, synergistic and even antagonistic effects that can result from such
480 cocktails as previously described (Khalil and Winder, 2008).

481 *4.2. Leachates cytotoxicity assessment*

482 Cytotoxicity analysis of the various leachate samples using single cell culture and serial dilution
483 of the samples (as measured by the MTS assay) was presented in Figure 3. The data in Figure 3
484 showed that the viability changed, as the leachates concentration increased from 0.313 to 80 %. In
485 sample A the viability significantly decreased after 2 h. from 95 to 67 % and after 24 h. from 78
486 to 9 %. Sample B the viability significantly decreased after 2 h. from 99 to 66 % and after 24 h.
487 from 86 to 22 %. Sample C viability also significantly decreased from 100 to 32 %, while sample
488 D viability decreased from 99 to 93 %. Sample E the viability decreased after 2 h. from 100 to 48
489 % and after 24 h. from 100 to 10 %. The data also indicated that 24 h. post leachates exposures we
490 witnessed statistically significant differences in damages between exposed and control cells (p
491 <0.05). The data clearly indicated the significant toxic potential of the undiluted leachates with
492 only one exception the leachates coming from Tripoli dumpsite post biological treatment which
493 proved less toxic than other leachates. This finding agrees with other researchers that investigated

494 contaminated soils from municipal dumps (Swati et al., 2017) and established loss of toxicity with
495 higher dilutions. The data also indicated a dose dependent significant increase in cell viability upon
496 dilution which could be correlated to dilution of chemical constituents in the leachate samples. The
497 observed cell death at higher leachates concentration could be the result of oxidative stress leading
498 to apoptosis and necrosis as previously reported by a number of researchers (Alabi et al., 2013)
499 (Talorete et al., 2008). Therefore the proposed cellular model using cell cultures and MTS
500 cytotoxicity assay (measuring cellular mitochondrial activity) (Mosmann, 1983) represented a
501 useful approach to measure and characterize landfill leachates toxicity with more sensitivity
502 achieved if cells are exposed 24 h. to the leachates prior to toxicological assessment. (Swati et al.,
503 2017) (Alimba et al., 2016).

504 *4.3. Leachates mutagenic potential*

505 The mutagenicity potential of chemicals identified in leachates was also reported by a number of
506 researchers using the comet assay methodology (Singh et al., 2007) (Noz et al., 1996; Collins,
507 2004; Deguchi et al., 2007; Gajski et al., 2012; Alimba et al., 2016). No literature could be found
508 on the genotoxic potential of municipal waste leachates from Lebanon. The results reported in
509 Table 4 consisted of calculated TMI. TMI is the tail DNA content of cells multiplied by the tail
510 length and divided by 1000.

511 The comet assay was used to establish the genotoxic potential of leachates samples as presented
512 in Table 3 and Figure 4. The tail moment TM analysis clearly indicated significant damages (p
513 <0.05) comparable to our positive control (KMnO_4) with the only exception being sample D the
514 biologically treated sample from Tripoli landfill where lower DNA damages were observed. This
515 agrees with other researchers that pointed out the important role of biological treatment and how

516 untreated leachates is characterized by higher genotoxicity by comparison to treated leachates
517 (Widziewicz et al., 2012) (Brkanac Sandra et al., 2014).

518 The cytotoxicity data together with the DNA damage data reported were used to rank the sites
519 from Most to least toxic (Table 4). The ranking of the site samples was similar although the DNA
520 damage assessment appeared to be more sensitive in detecting early damage levels.

521

522 **Table 4:** Leachates sample ranking from most to least toxic using the MTS and Comet assays.

MTS assay results ranking 24 h. post exposure ranking (based on LD₅₀) (Most toxic to least toxic)	Comet Assay Tail DNA Content Ranking (Most to least damage)
1. Sample B	1. Sample A
2. Sample A	2. Sample B
3. Sample E	3. Positive control (KMnO ₄)
4. Sample C	4. Sample E
5. Sample C	5. Sample C
6. Sample D	6. Sample D
7. Negative control (HBSS)	7. Negative control (HBSS)

523

524

525

526 *4.4. Leachates assessment using the mouse model*

527 The cytotoxicity data presented in this paper clearly indicated significant levels of cellular and
528 DNA damages in cells exposed to leachates. Unfortunately the generated toxicity profiles reported
529 using cellular systems does not always translate to animal model as reported by a number of
530 researchers (Jakasa and Kezic, 2008) (Nesslany, 2017). Therefore we selected an animal model
531 the BALB/c mice model as a biological indicator as previously reported by many researchers
532 (Farombi et al., 2012) (Bakare et al., 2003) to assess the impacts of leachates exposure *in vivo* and
533 to further validate the *in vitro* results.

534 The mice drinking water was substituted with diluted and pure leachates for a period of 4 weeks
535 and data reported in Figure 5. The average organ weight of the kidneys in samples A, B, C and D
536 was 0.17 grams while in controls 0.15 grams, that of the liver in the four samples between 0.76
537 and 0.93 grams while in controls 0.81 grams, and that of the spleen in the four samples between
538 0.11 and 0.15 grams while in controls 0.18 grams. Therefore we concluded that no significant
539 differences in the weight of the mice between exposed and controls could be observed.
540 Furthermore, kidney spleen and liver harvested from euthanized mice at the end of the exposure
541 period did not show any significant differences in weight by comparison to controls. These
542 observations were not in agreement with other studies on mice (Bakare et al., 2003) where it was
543 determined that weight loss could be observed after leachates exposure. This discrepancy may be
544 attributed to the subtle differences in leachate composition between landfill sites.

545 The enzyme activities averages in samples A, B, C and D were measured. ALP of liver ranged
546 between 100 and 257 %, the ALP of blood between 109 and 160 %, the GPT of liver between 62
547 and 98 %, the GPT of blood between 89 and 100 %, the GOT of liver between 102 and 160 %,
548 and the GOT of blood between 514 and 772 %. We reported significant increases in ALP levels in
549 the livers (homogenized liver samples) upon leachates exposures in sample B, C and D. We also
550 reported significant increases in liver GOT levels in samples A and B. No significant changes in
551 GPT levels could be reported in all the samples. The observed significant expression of GOT and
552 ALP levels in the livers of exposed mice is a clear indication of changes in liver function because
553 activity of this enzyme has been reported as a good diagnostic tool for the damages in human liver
554 (Fernandes et al., 2011) (Limdi and Hyde, 2003). The level of enzymatic activity was also
555 measured in the mice blood (ALP, GOT and GPT) (Figure 6). The data indicated significant
556 expression of GOT levels in all samples. This is a very clear indication of acute liver damage being

557 triggered by the leachates exposure over 4 weeks. This elevation in blood levels of GOT indicates
558 a possible disease of the heart, liver or muscle (Babb, 1973).
559 Further investigation of leachates toxicity in mice was also undertaken at the DNA levels
560 measuring the DNA damages in isolated cells from various organs of the exposed mice (Figure 7).
561 The data presented in Figure 7 clearly indicated significant DNA damages in all cells isolated from
562 these organs exposed to all the leachate samples. This is in agreement with many publications
563 correlating DNA damage in bone marrow and blood in exposed mice to leachates in other countries
564 (Siddique et al., 2005; Chandra et al., 2006) (Tewari et al., 2006).

565

566 **5- Conclusion**

567 This study investigated the potential risks posed by municipal waste leachates in Lebanon using
568 GC/MS, chemical analysis and toxicological testing. The fate of the majority of leachates was the
569 Mediterranean Sea. The findings of this research clearly indicated the presence of numerous
570 organic compounds such as oleamide, hentriacontane, eicosane, tributylamine, phthalic acid
571 derivatives, nicotine and pentacosane. The presence of DEHP in almost all samples along with
572 other organic compounds can pose potential health risk to humans and aquatic species. This finding
573 together with high levels of nitrates, phosphates, heavy metals and high levels of Mn, Cr, Ni, Cd
574 and others is very alarming given the lack of efficient biological leachate treatment methods in
575 Lebanon. Some of these identified substances posed significant health and carcinogenic risks and
576 the levels measured were in most cases way above acceptable levels in published drinking water
577 and US EPA guidelines. The leachates biological toxicity analysis using *in vitro* and *in vivo* assays
578 showed the hazardous nature of the leachates especially in undiluted form in triggering acute
579 toxicity and reactions in exposed cells and mice. Leachates were shown to decimate cell viability

580 in addition to triggering significant DNA damages in human derived cell cultures and mice cells.
581 The *in vivo* data also indicated significant short-term damages in the blood and organs such as the
582 liver which was coupled with significant damages at the DNA levels in exposed animals.
583 These findings clearly indicated the high public health risks posed by the leachates and the urgent
584 need to establish better national management of biological treatment and disposal of these
585 leachates. The *in vitro* and *in vivo* data also agreed on the significant damages triggered at the
586 molecular levels which can lead to cancer onset upon exposure.

587

588 **5- Conflict of interest**

589 The authors declare that here is no conflict of interest regarding the publication of this paper.

590

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